

**FACULTY OF ENGINEERING & INFORMATICS****B.E. I-Semester (Suppl.) Examination, June / July 2017****Subject : Engineering Chemistry-I****Time : 3 hours****Max. Marks : 70****Note: Answer all questions from Part-A. Answer any FIVE questions from Part-B.****PART – A (20 Marks)**

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| 1  | State and explain First law of thermodynamics.   | 2 |
| 2  | State Carnot theorem.  | 2 |
| 3  | What is meant by the term eutectic?  | 2 |
| 4  | State phase rule.  | 2 |
| 5  | Calculate the carbonate and non-carbonate hardness of a sample of water in ppm containing : $\text{Ca}(\text{HCO}_3)_2 = 8.1 \text{ mg.L}^{-1}$ ; $\text{Mg}(\text{HCO}_3)_2 = 7.3 \text{ mg.L}^{-1}$ ; $\text{MgCl}_2 = 9.5 \text{ mg.L}^{-1}$ ; $\text{CaSO}_4 = 13.6 \text{ mg.L}^{-1}$ . | 2 |
| 6  | Define the terms i) Scale and ii) sludge   | 2 |
| 7  | Give one example each for Addition and Condensation polymers.  | 2 |
| 8  | Write the structures of poly-acetylene and poly-aniline.   | 2 |
| 9  | Define the terms i) Saponification number and ii) acid value   | 2 |
| 10 | Explain the property of RUL in refractories.   | 2 |

**PART – B (50 Marks)**

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| 11 a) | Calculate the maximum work done when 2 moles of an ideal gas expand isothermally and reversibly from a volume of 10 litres to a volume of 20 litres at 298k. | 5 |
| b)    | Explain the criteria for spontaneity of a process in terms of entropy and free energy.   | 5 |
| 12 a) | What do you understand by the reduced phase rule equation? Discuss the use of the phase rule in Pattinson's process of desilverization of lead.              | 5 |
| b)    | Define the terms i) Phase ii) component iii) degrees of freedom.   | 5 |
| 13 a) | Explain the procedure for the determination of Alkalinity of water.  | 5 |
| b)    | Discuss the concept of break point chlorination.   | 5 |
| 14 a) | Explain the preparation, properties and applications of Nylon-6,6.   | 5 |
| b)    | Differentiate between thermoplastic and thermosetting polymers.  | 5 |
| 15 a) | Classify lubricants and give one example each for various type of lubricants.  | 5 |
| b)    | Write a note on the following properties of Refractories.<br>i) Refractoriness ii) Thermal spalling  | 5 |
| 16 a) | Derive an expression for the efficiency of heat engine by using carnot cycle.  | 6 |
| b)    | Discuss the ion-exchange method of softening hard water.   | 4 |
| 17 a) | Write a note on intrinsic conducting polymers  | 5 |
| b)    | Explain the terms i) viscosity index ii) glazing   | 5 |